

# Iron Assay Kit (Fe)

Method: Ferrozine

Cat .No.	Size	Instrument	
GB440E	R1: 4×40 ml	For Hitachi 717	
GD440E	R2: 2×20 ml	& ShimadzuCL7200/8000	
GS441E	R1: 4×40 ml	For Hitachi917	
GS441E	R2: 2×20 ml	& OlympusAU640/400/600	
GX441F	R1: 1×80 ml	For SYNCHRON CX4-5-7-	
GX441E	R2: 1×20 ml	9/LX20/DXC600-800	
GT441E	R1:4×40 ml	For TOSHIBA	
	R2:2×20 ml		

## **INTENDED USE**

For the quantitative *in vitro* determination of Serum Iron in serum or plasma.

# CLINICAL SIGNIFICANCE [1, 2]

Iron measurements are used in the diagnosis and treatment of diseases such as iron deficiency anaemia, hemochromatosis (a disease associated with widespread deposit in the tissues of two iron-containing pigments, hemosiderin and hemofuscin, and characterized by pigmentation of the skin), and chronic renal disease.

## **ASSAY PRINCIPLE**

Ferric iron is dissociated from its carrier protein, transferrin, in an acid medium and simultaneously reduced to the ferrous form. The ferrous iron is then complexed with the chromogen, a sensitive iron indicator, to produce a coloured chromophore which absorbs at 560 nm.

# SAMPLE COLLECTION AND PREPARATION

Serum, heparinized plasma. Haemolysis and plasma treated with oxalate interfere with the test. Even minimal concentrations of EDTA in the sample lead to depressed results.

The samples stable for 7 days at 2-8  $^{\circ}$ C, or for 4 days at 15-25  $^{\circ}$ C.

# REAGENT COMPOSITION

Contents	Concentration of Solutions	
Acetate Buffer	100 mmol/L, pH 4.2	
Ferrozine	5 mmol/L	
Deoxidizer and Surfactant		

# STABILITY AND PREPARATION OF REAGENTS

All reagents are ready to use.

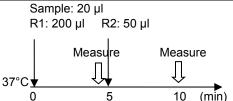
Stable up to the expiry date when stored at 2-8°C.

# **ASSAY PROCEDURE**

**Test Procedure for Analyzers** 

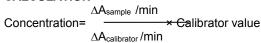
Assay Mode: 2 Point End

Wave Length (main/sub): 570 nm/700 nm



- Mix 20 µl sample with 200 µl R1 and incubate at 37°C for 5 minutes, then read initial absorbance A₁.
- Add 50 μl R2 into cuvette, mix and incubate for 5 minutes at 37°C, Read final absorbance A<sub>2</sub>.
- Calculate the absorbance change ΔA=A<sub>2</sub>-A<sub>1</sub>.

#### **CALCULATION**



#### **CALIBRATION**

Recommend that this assay should be calibrated using Randox Calibration Serum Level 3 or Level 2.

# **QUALITY CONTROL**

Randox Assayed Multisera, Level 2 and Level 3 are recommended for daily quality control. Two levels of controls should be assayed at least once a day. Values obtained should fall within a specified range. If these values fall outside the range and repetition excludes error, the following steps should be taken:

- 1. Check instrument settings and light source.
- 2. Check reaction temperature.
- 3. Check expiration date of kit and contents.

#### **NORMAL VALUE**

NORMAL VALUE	ug/dl	umol/L
Men	59-158	10.6-28.3
Women	37-145	6.6-26.0

It is recommended that each laboratory establish its own reference range to reflect the age, sex, diet and geographical location of the population.

# **UNIT CONVERSION**

 $\mu$ g/dl × 0.179 =  $\mu$ mol/L

# SPECIFIC PERFORMANCE CHARACTERISTICS

## **LINEARITY**

The method is linear up to 180  $\mu$ mol/L. If the sample above this concentration should be diluted with 0.9% NaCl and repeat assay. Multiply the result by dilution factor.

## **PRECISION**

The CV of the test should be less than 5%

Intra assay precision				
N=20	Level1	Level 2		
Mean (µmol/L)	18.58	36.07		
SD	0.12	0.34		
CV	0.63%	0.95%		
Inter assay precision				

Beijing Strong Biotechnologies, Inc.

Add:5/F Kuang Yi Building, No. 15 Hua Yuan Dong Lu, Haidian District, Beijing 100191 P. R. China

Tel: +86 10 8201 2486 Fax: +86 10 8201 2812





N=5	Level1	Level 2
Mean (µmol/L)	18.27	35.73
SD	0.25	0.39
CV	1.35%	1.09%

EC REP Authorized Representative in the **European Company** 

## **SENSITIVITY**

The minimum detectable level that can be distinguished from zero has been determined as 1.11 µmol/L.

#### **INTERFERENCE**

The following analyze were tested up to the levels indicated and found not to interfere:

Hemoglobin: 250 mg/L Intralipid: 500 mg/L Bilirubin: 600 mg/L Ascorbic Acid: 500 mg/L Cu<sup>2+</sup>: 50 µmol/L Zn<sup>2+</sup>: 80 umol/L

# **SAFETY PRECAUTIONS AND WARNINGS**

- For in vitro diagnostic use only. Do not pipette by mouth. Exercise the normal precautions required for handing laboratory reagents.
- The reagents contains Sodium Azide. Avoid ingestion or contact with skin or mucous membranes. In case of skin contact, flush affected area with copious amounts of water. In case of contact with eyes or if ingested, seek immediate medical attention.
- Sodium Azide reacts with lead and copper plumbing, to form potentially explosive azides. When disposing of such reagents flush with large volumes of water to prevent azide build up. Exposed metal surfaces should be cleaned with 10% sodium hydroxide.
- All specimens used in this test should be considered potentially infectious. Universal Precautions, as they apply at your facility, should be used for handling and disposing of materials during and after testing.

# **REFERENCES**

- Ceriotti, F., and Ceriotti, G., Improved Direct Specific. Determination of Serum Iron. Clin. Chem. 26/2, 327-331 (1980).
- Henry, R.J.: Clinical Chemistry Principles and Techniques, New York, Harper and Row (1968) pg.386.

#### **INDEX OF SYMBOLS**

REF

Manufacture

LOT

Catalogue Number Lot number

Date of manufacture



Use by(Expiration date)



For In-Vitro Diagnostic use only



Stored at 2-8°C



Attention:See instruction for use

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Add:5/F Kuang Yi Building, No. 15 Hua Yuan Dong Lu, Haidian District, Beijing 100191 P. R. China

Tel: +86 10 8201 2486 Fax: +86 10 8201 2812



Web: www.bsbe.com.cn Email: jq.tech@bsbe.com.cn