

Copper Assay Kit (Cu)

Method: PAESA Chromogenic Agent

Cat .No.	Size	Instrument		
GB9450E	R1: 1×60 ml	For Hitachi 717		
	RZ. 1^20 IIII	& ShimauzuCL/200/6000		
GS9451E	R1: 1×60 ml	For Hitachi 917		
	R2: 1×20 ml	& OlympusAU640/400/600		

INTENDED USE

For the *in vitro* quantitative determination of Copper in serum.

CLINICAL SIGNIFICANCE

The assay kit is for determination of Copper (Cu). As one of the most important trace elements, Copper(Cu) is a part of many metal enzymes and participate in synthesizing melanin, collagen.

The decrease of copper may cause hypogenesis, anaemia of sex of cellule low pigment. The acute toxicity of copper can cause acute renal failure and gastro-enteritis.

ASSAY PRINCIPLE^[1,2]

CER-Cu²⁺ <u>Acidic conditions</u> CER+Cu²⁺

Albumin-Cu²⁺ <u>Acidic conditions</u> Albumin+Cu²⁺

 $Cu^{2+}+VC(reduced) \longrightarrow Cu^{+}+VC(oxidized)$

Cu⁺+3,5-DiBr-PAESA → Complex (blue)

Calculate the copper concectration by the adsorbance at 600 nm.

SAMPLE COLLECTION AND PREPARATION

Serum samples.

Serum samples are stable for a week at 2-8°C,

REAGENT COMPOSITON

Contents	Concentration of Solutions	
Reagent 1 (R1) Buffer		
Vc(reduced)		
Reagent 2 (R2)		
3.5-Di-Br-PAESA	0.1mmol/L	

STABILITY AND PREPARATION OF REAGENTS All reagents are ready to use.

Stable up to the expiry date when stored at 2-8°C. The Cu assay kit reagents are stable for 30 days on board.

ASSAY PROCEDURE

Wave Length (main/sub): 600 nm/700 nm

Sample: 10 μl R1: 150 μl R2: 50 μl



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- 1. Mix 10 μ I sample with 150 μ I R1 and incubate at 37 °C for 5 minutes, then read initial absorbance A₁.
- Add 50 μl R2 into cuvette, mix and incubate for 5 minutes at 37°C, Read final absorbance A₂.
- 3. Calculate the absorbance change $\Delta A = A_2 A_1$.

CALCULATION

△A_{sample} /min Concentration=

 $\Delta A_{calibrator}/min$

CALIBRATION

Recommend that this assay should be calibrated using Randox complex Calibrator or Gcell Cu Calibrator.

Calibrator value

QUALITY CONTROL

For quality control, use Randox complex Control as daily quality control sera and can be purchased separately. Values should fall within a specific range. If these values fall outside the range and repetition excludes error, the following steps should be taken:

- 1. Check instrument settings and light source.
- 2. Check reaction temperature.
- 3. Check expiration date of kit and contents.
- 4. Check the quality of the water used for reagents reconstitution.

REFERENCE VALUE

UNIT CONVERSION

µg/dl × 0.1574=µmol/L

SPECIFIC PERFORMANCE CHARACTERISTICS

LINEARITY

The method is linear up to 100μ mol/L. Samples above this concentration should be diluted 1+1 with 0.9% NaCl and reassay. Multiply the result by 2.

PRECISION

The CV of the test should be less than 10%

Intra assay precision				
N=20	Level1	Level 2		
Mean (µmol/L)	17.18	28.87		
SD	0.25	0.33		
CV	1.48%	1.14%		
Inter assay precision				
N=5	Level1	Level 2		
N=5 Mean (µmol/L)	Level1 17.84	Level 2 29.70		
N=5 Mean (µmol/L) SD	Level1 17.84 0.43	Level 2 29.70 0.40		

SENSITIVITY

The minimum deteccttable concentration of $\ Cu^{2+}$ with an acceptable level of precision was determined as 1.97 $\mu mol/L.$

INTERFERENCE

A Reagent blank may be performed by replacing sample or standard with double deionized water. The following





analyze were tested up to the levels indicated and found
not to interfere:Hemoglobin:100 mg/dlIntralipid:500 mg/dlBilirubin:100 mg/dlUric Acid:250 mg/dlD-penicillamine :250 mg/dl

200 mg/dl

Sodium heparin : CORRELATION

This method (Y) was compared with another commercially available method (X) and the following linear regression equation obtained:

Y=1.0513X+0.2096, and a correlation coefficient of 0.9159, 48 patient samples were analyzed.

SAFETY PRECAUTIONS AND WARNINGS

- 1. For in vitro diagnostic use only. Do not pipette by mouth. Exercise the normal precautions required for handing laboratory reagents.
- Solution contains Sodium Azide. Avoid ingestion or contact with skin or mucous membranes. In case of skin contact, flush affected area with copious amounts of water. In case of contact with eyes or if ingested, seek immediate medical attention.
- All specimens used in this test should be considered potentially infectious. Universal Precautions, as they apply at your facility, should be used for handling and disposing of materials during and after testing.

REFERENCES

- Abc,S.Yamashita, A. Noma, Sensitive, direct colorimetric assay for copper in serum, Clin Chem, 35,552-554(1989)
- Katarzyna Zawistowska. Copper chelate with 2pyridylazo ligands as test probes for characterization of micellar effects. COLLOIDS AND SURFACES, 2008, 315: 259-267

INDEX OF SYMBOLS



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